

Theoretical Chemistry at UNICAMP Molecular Dynamics / Skaf Group

Biophysical Molecular Dynamics Simulations

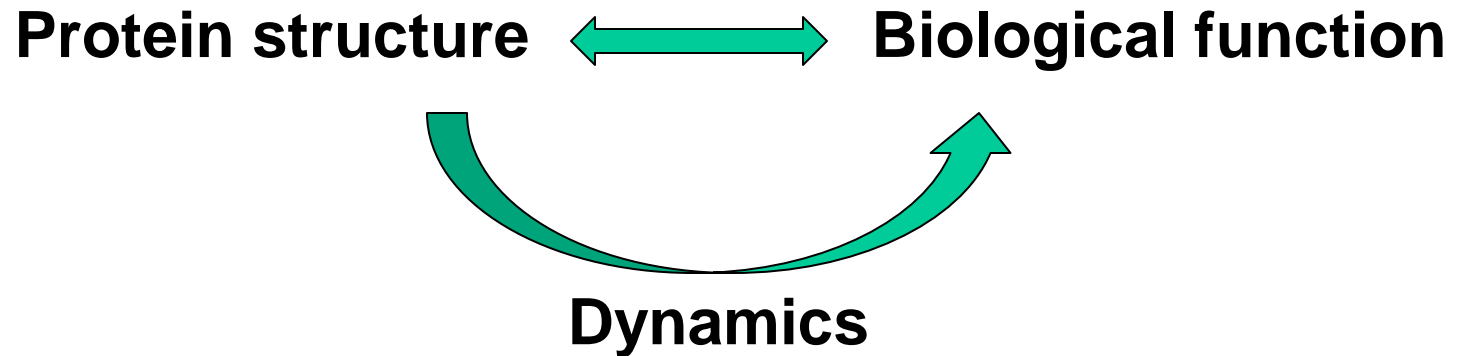
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Institute of Chemistry / State University of Campinas

BIOEN Workshop on Molecular Mechanisms of Photosynthesis

São Paulo, October 25, 2010

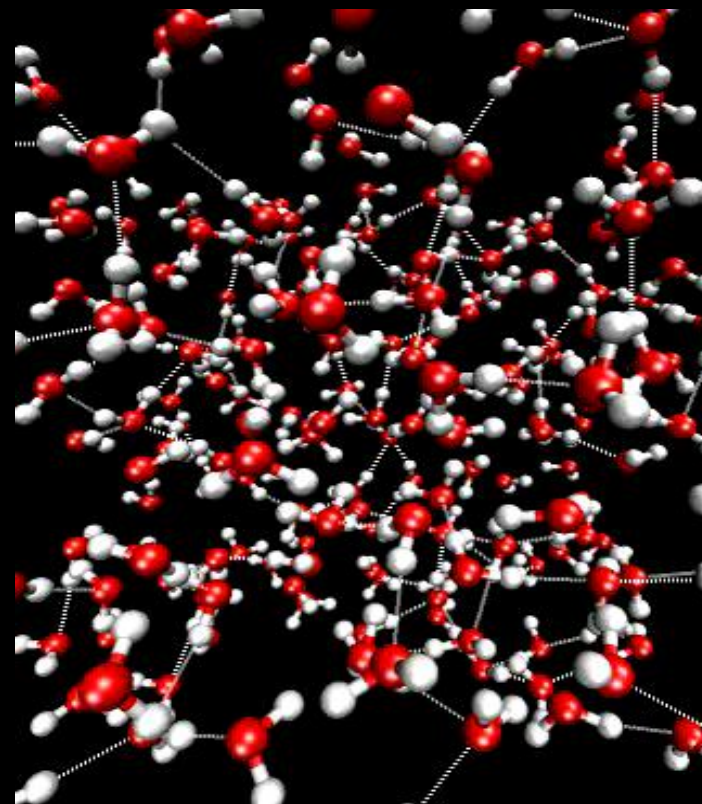
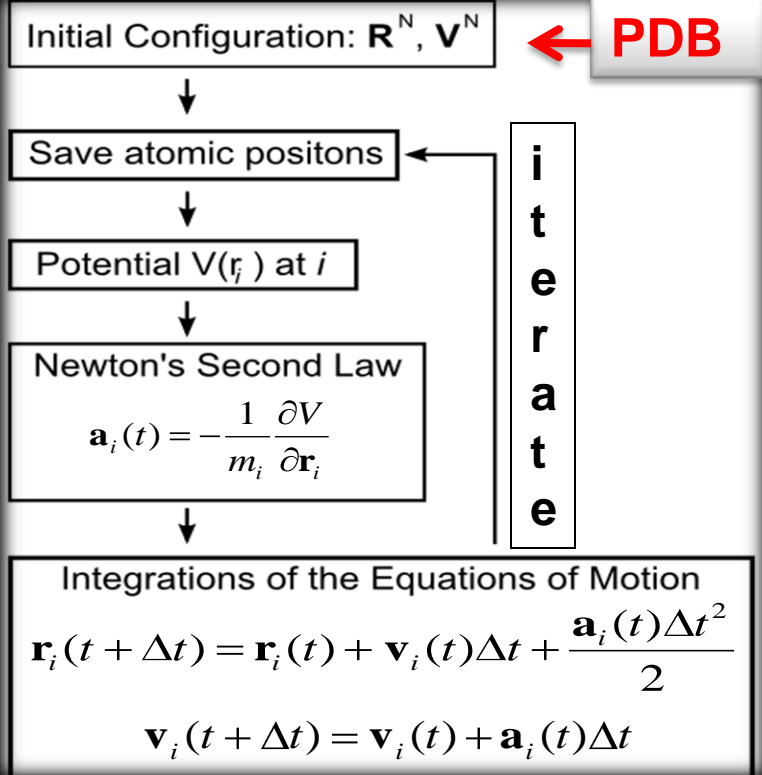
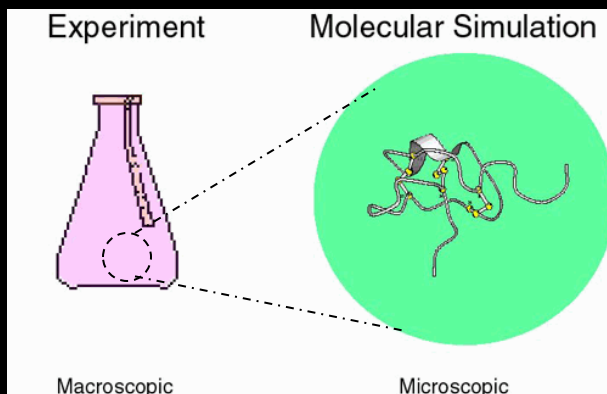
The Holy Grail of Structural Molecular Biology



Understanding how molecules move requires knowledge of interactions at the atomic level

MD simulations are a powerful tool for studying atomic motions

MD Simulations in a Nutshell

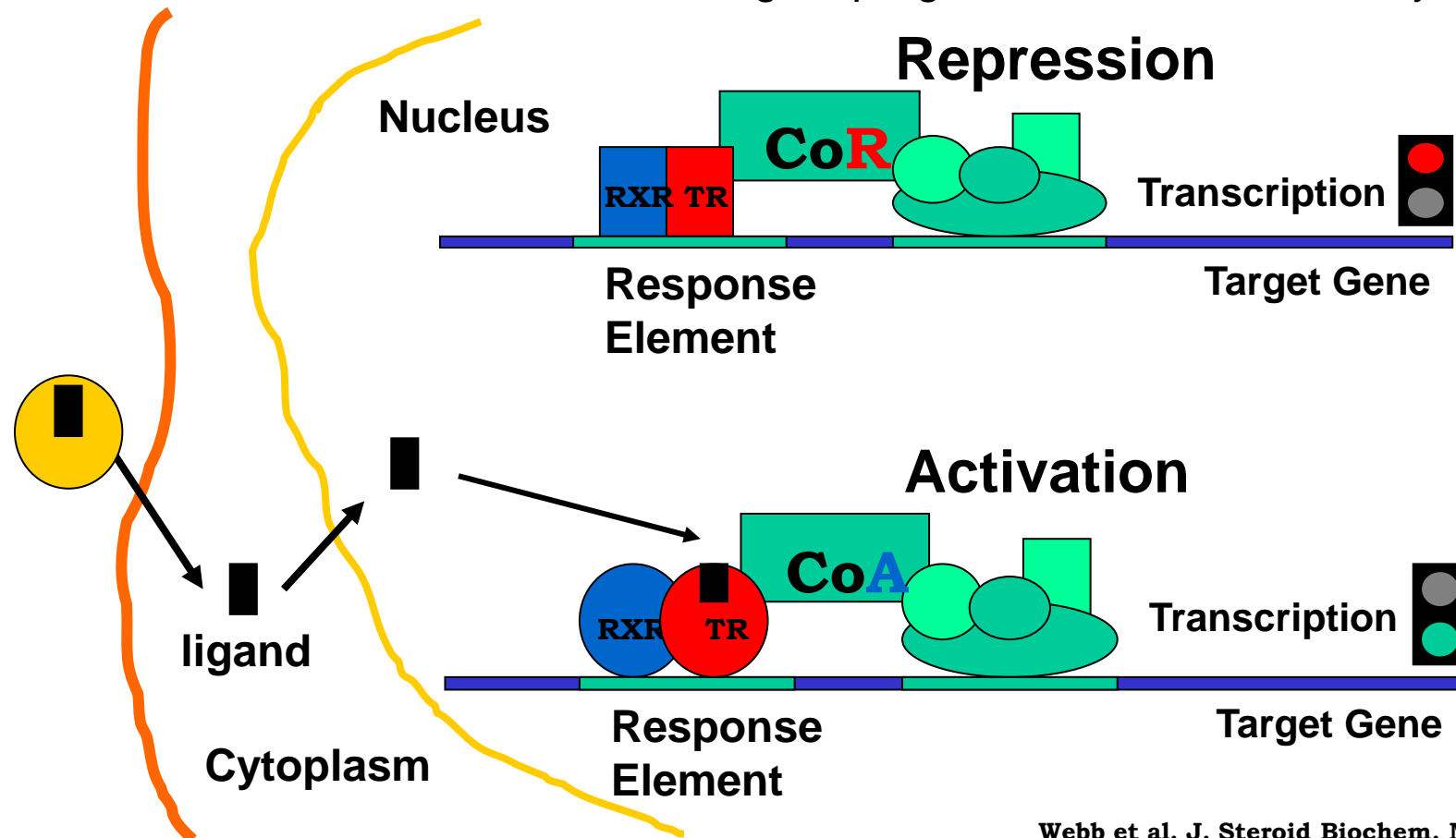


Nuclear Receptors (transcription proteins)

Role: Modulate gene expression by means of hormone binding

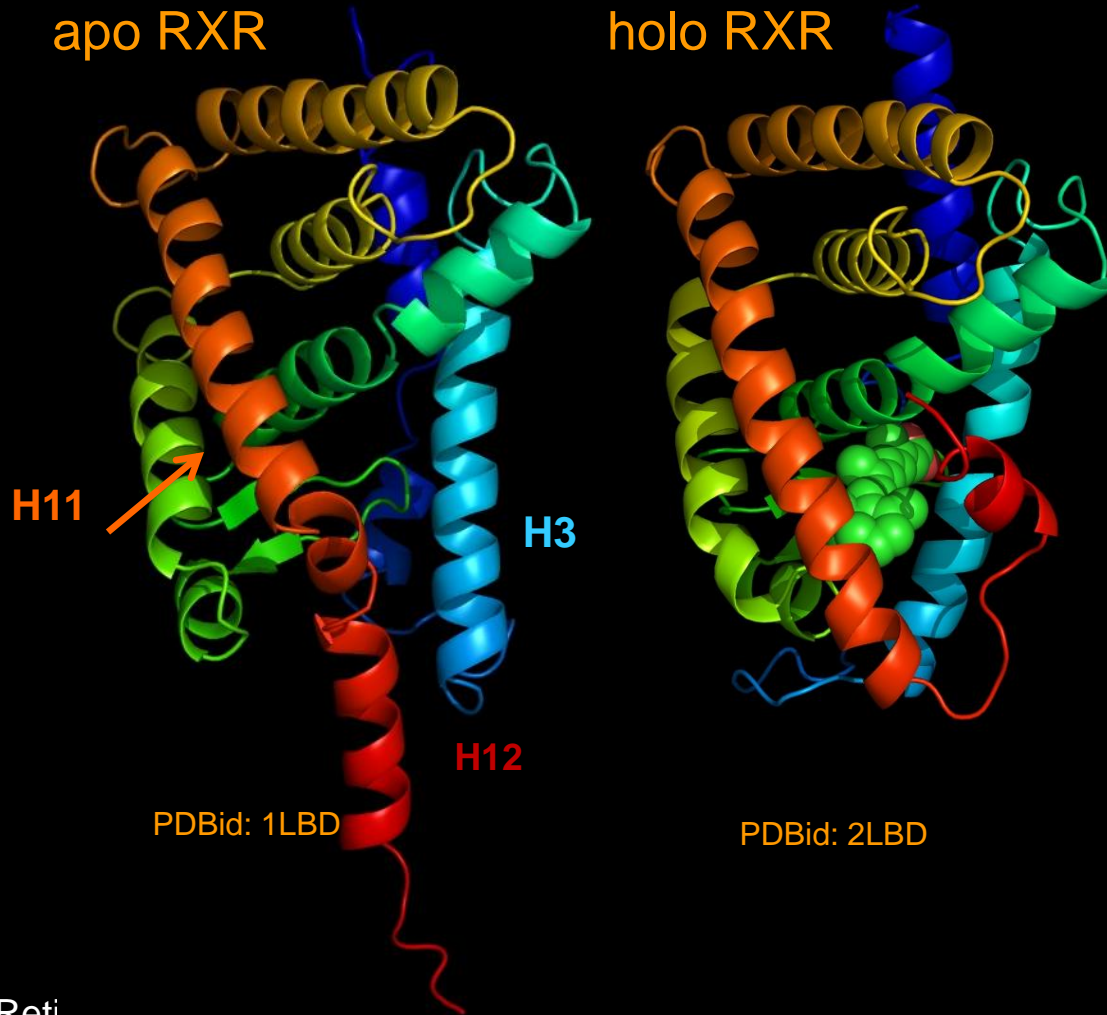
Importance: Cell differentiation, metabolism, sexual diff. & function, diabetes, cancer, inflammatory processes, etc

Vitamins A&D, steroids, estrogen, progesterone, corticoids, fatty acids, etc



HOW DO LIGANDS ENTER & EXIT NR'S HYDROPHOBIC BINDING POCKET ?

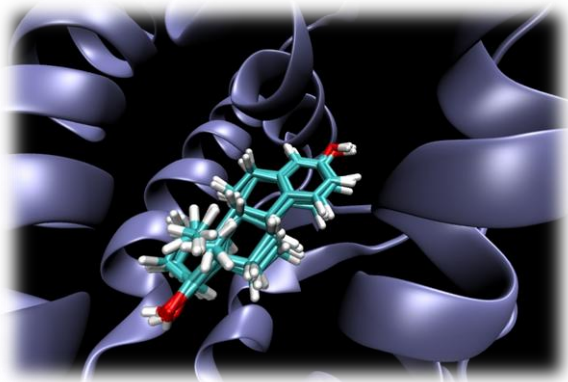
The Mouse Trap Model



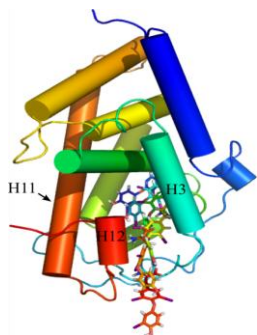
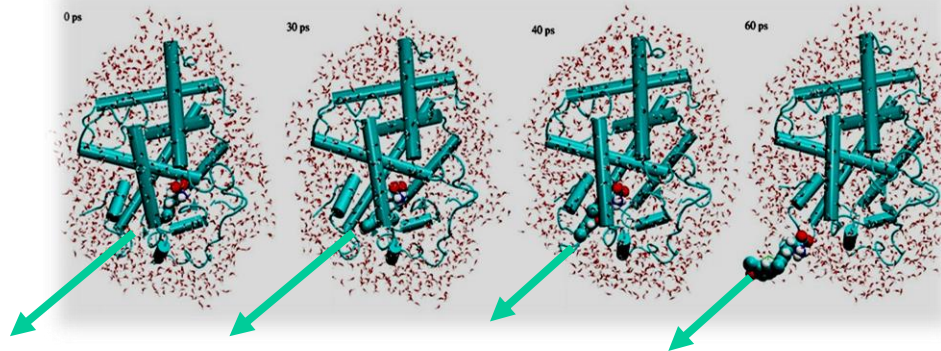
RXR: 9-*cis*-Reti

Mechanisms of Ligand Dissociation

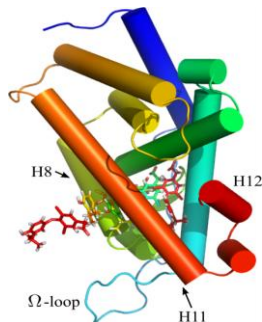
**Locally Enhanced Sampling:
Search for dissociation pathways**



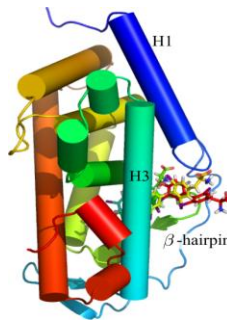
**Steered MD:
Relative importance of pathways**



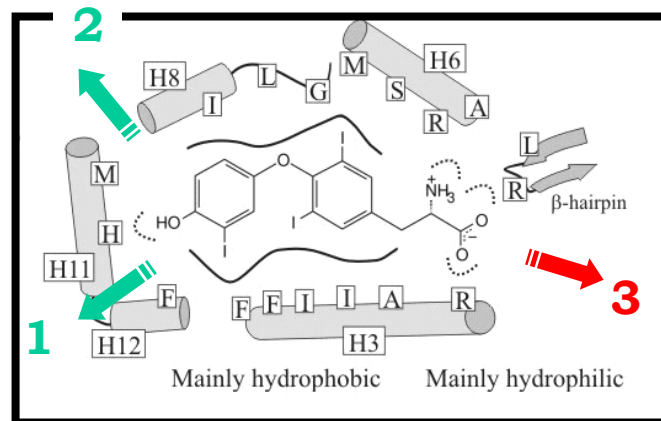
Path I



Path II



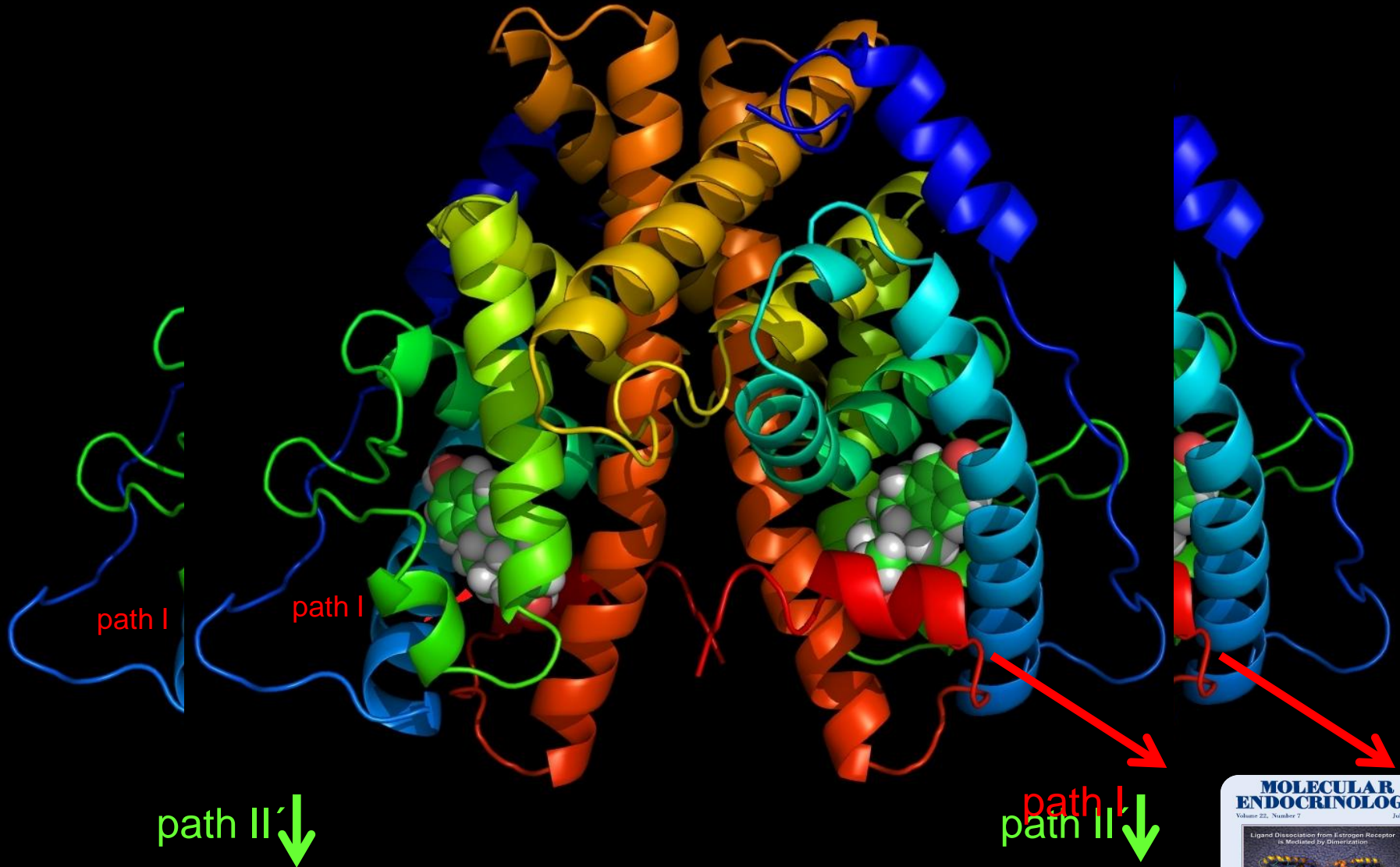
Path III



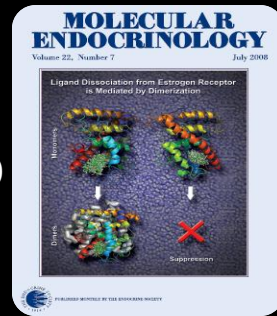
Path 3: Ligand-protein hydrophilic contacts are readily replaced by ligand-solvent interactions

NR can harbor multiple ligand escape pathways

Dissociation Pathways from Estrogen Receptor Dimers



Dimerization strongly suppresses some dissociation paths. MD suggests explanation for ligand dissociation kinetics in ER.

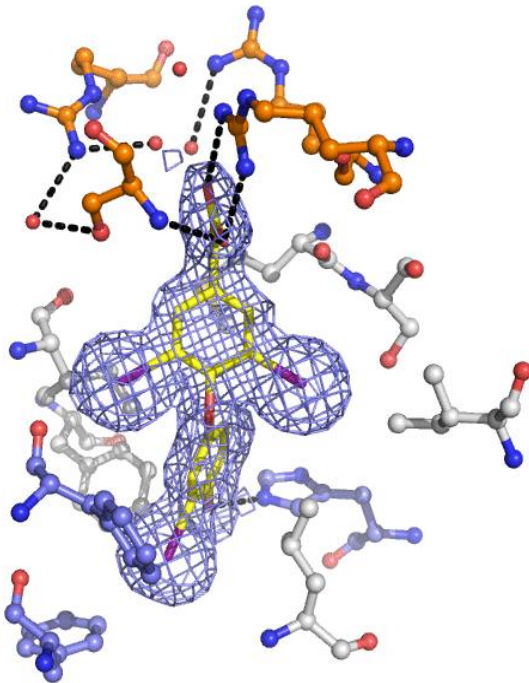


Thyroid Hormone (TR) Isoform Selectivity

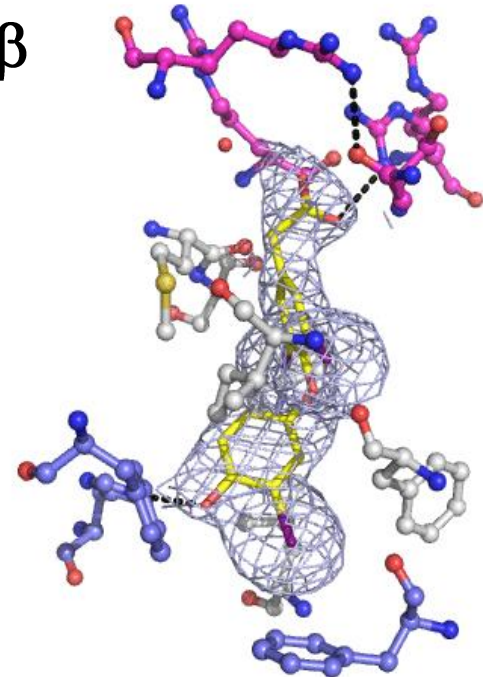
TR α : Abundant in the heart / TR β : Abundant in the liver

TRIAC is a natural ligand that binds TR β with higher affinity.

TR α



TR β



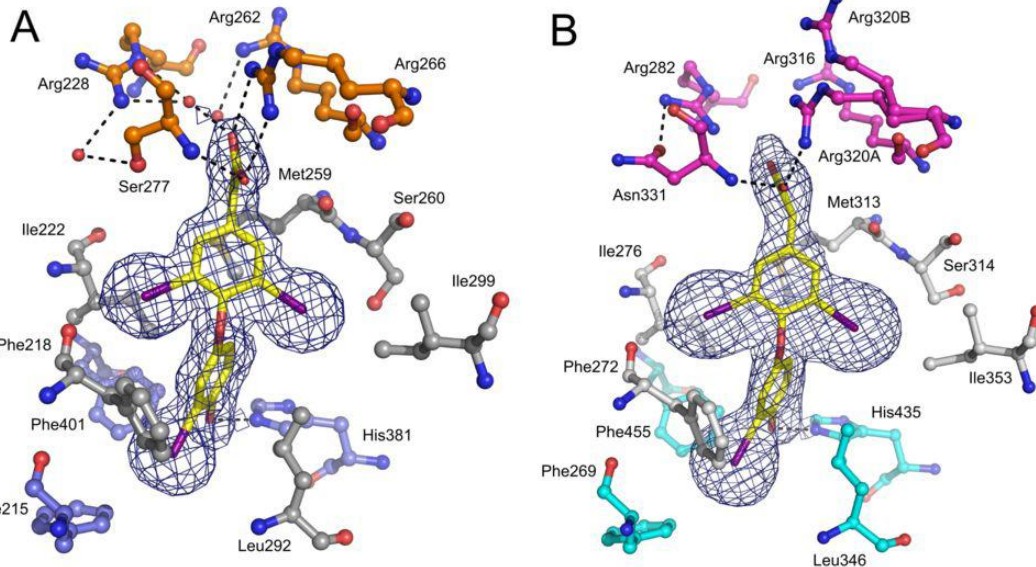
Crystal structures

Thyroid Hormone (TR) Isoform Selectivity

TR α : Abundant in the heart / TR β : Abundant in the liver

TRIAC is a natural ligand that binds TR β with higher affinity.

Apparent paradox: X-ray structures clearly indicate stronger Triac binding to TR α , not to TR β !



Crystal structures

MD

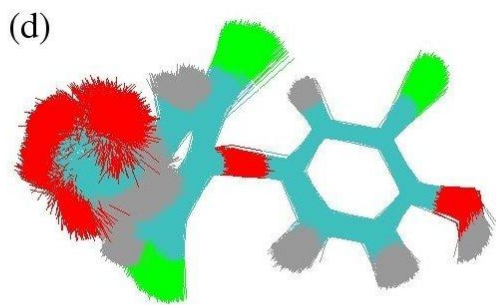
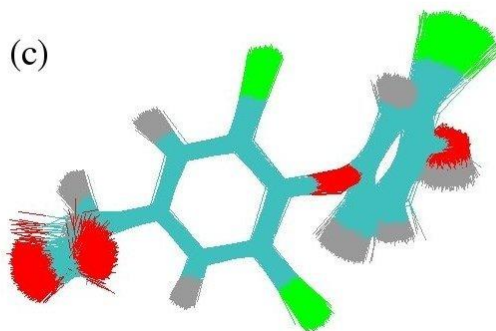
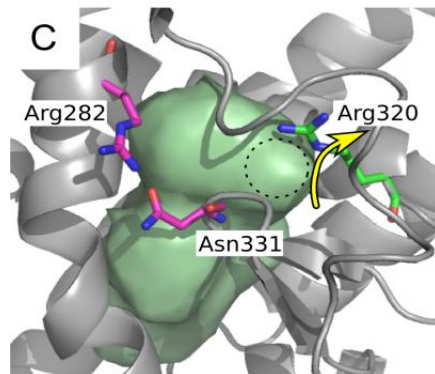
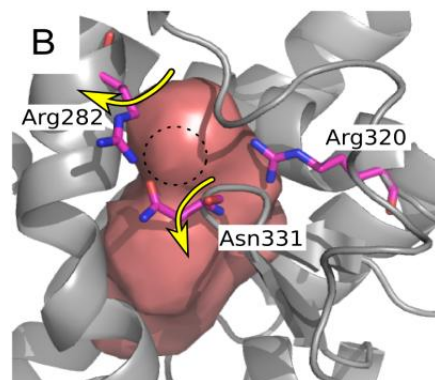
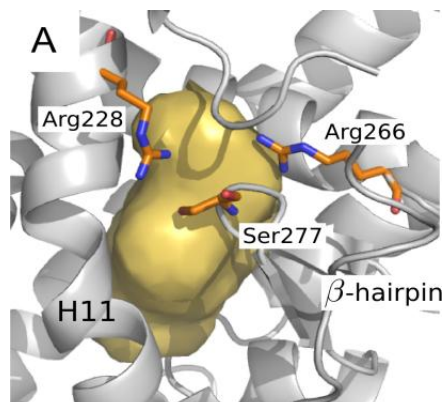
Ligand interactions and the role of water

TRIAC Interaction with:	TR α / kcal mol $^{-1}$	TR β / kcal mol $^{-1}$
LBD residues	-66.63	-46.96
Whole environment	-192.74	-192.52
Water molecules	-18.37	-48.47

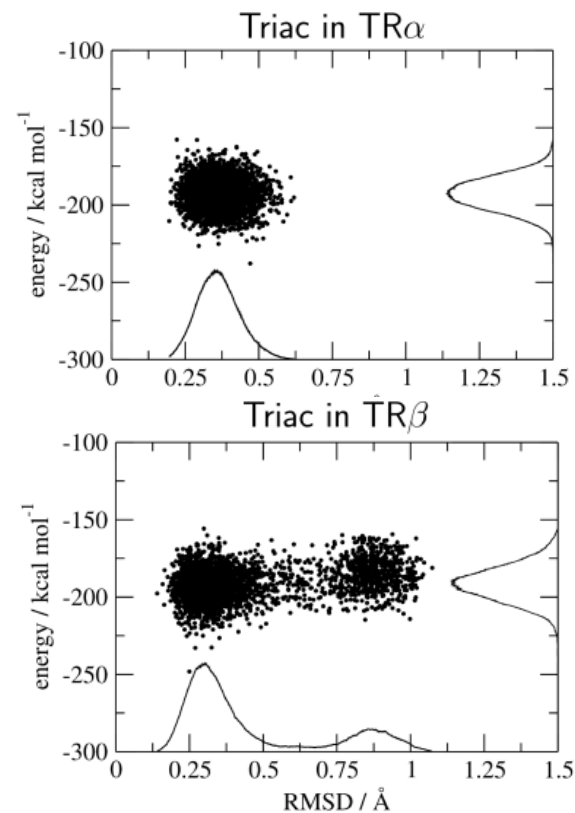
- Ligand-LBD interactions with TR α are, indeed, stronger!
- Interactions with the environment are similar...
- Water substitutes Ligand-LBD interactions!

Triac/TR α : 1 water at a time
Triac/TR β : up to 4 waters in cavity

TRiAC is substantially more mobile in TR- β cavity



Computing entropy difference



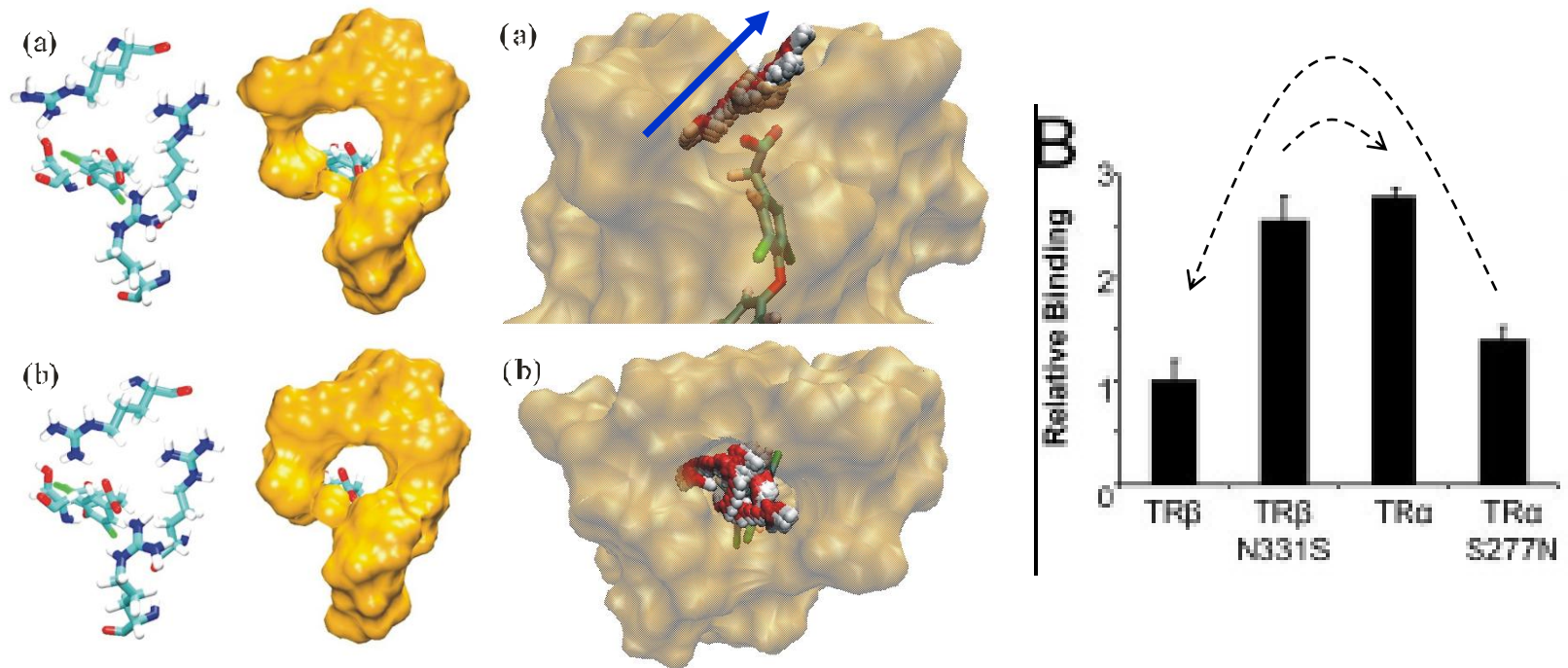
$$\Delta S = R \ln (\Omega_B / \Omega_A)$$

MD: Entropy accounts for Triac selectivity

MD: ~ 0.4 kcal/mol

Exp: ~ 0.4-0.6 kcal/mol

Further analyses of binding pocket suggest that specific S277/N331 mutation should lead to near reversal of TR α /TR β selectivity.



Functional studies confirms MD hypothesis

Martínez et al. PNAS 106, 20717 (2009)

Gaining ligand selectivity in thyroid hormone receptors via entropy

Leandro Martínez¹, Alessandro S. Nascimento^{1,4}, Fabio M. Nunes¹, Kevin Phillips¹, Ricardo Aparício¹, Sandra Maria G. Dias^{1,2}, Ana Carolina M. Figueira³, Juan H. Lin¹, Phuong Nguyen¹, James W. Apriletti¹, Francisco A. E. Newer¹, John D. Storde^{1,4}, Paul Webb^{1,4}, Munir S. Khan^{1,4}, and Igor Poljakov^{1,4}

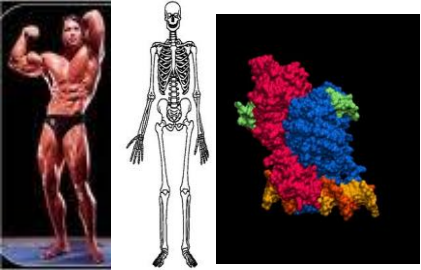
¹Instituto de Química, Universidade Federal de Campinas, 13160-970, Campinas, Brazil; ²Instituto de Física de São Carlos, Departamento de Física e Informática, Universidade de São Carlos, 13560-970, São Carlos, Brazil; ³Medicinal Hospital Research Institute, Houston, TX 77030; and ⁴Universidade de Pernambuco, Faculdade de Ciências da Saúde, Universidade de Brasília, Campus Universitário Darcy Ribeiro, 07190-900, Brasília, Brazil

Contributed by John D. Storde, October 12, 2009 (first published September 8, 2009)

PNAS

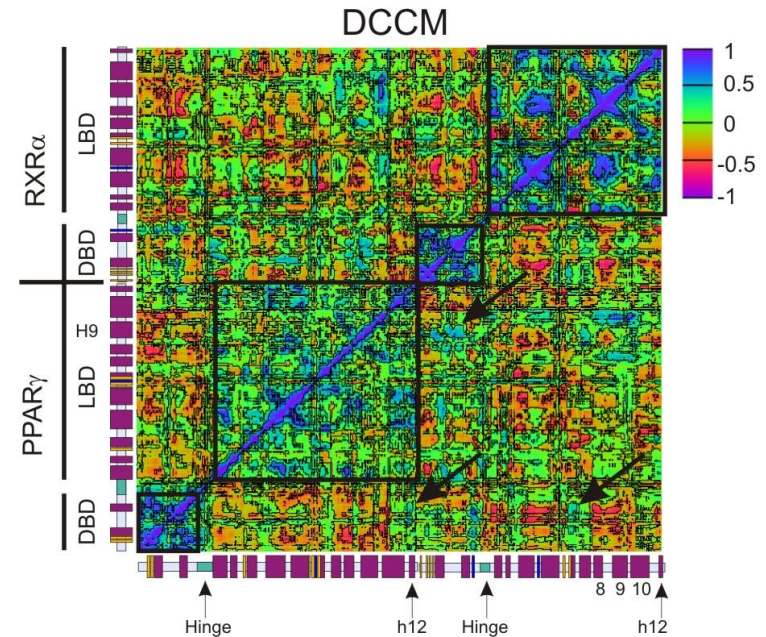
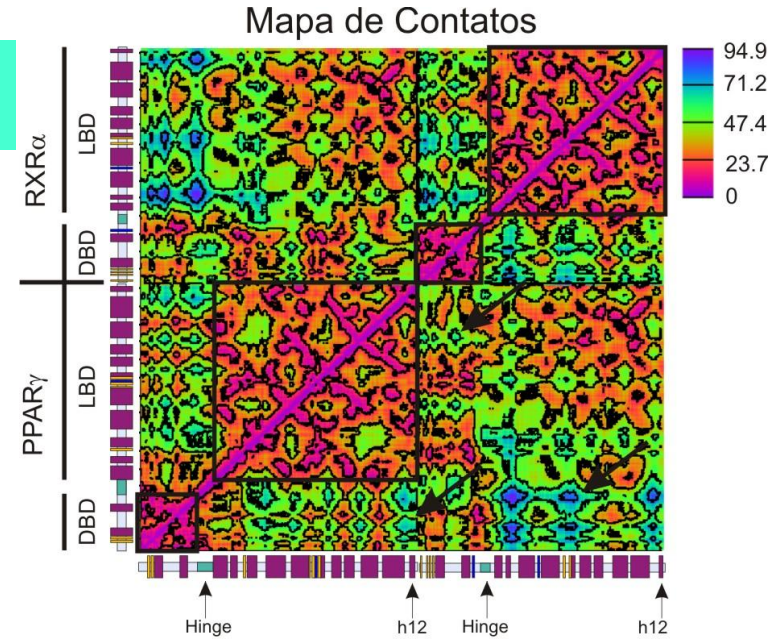
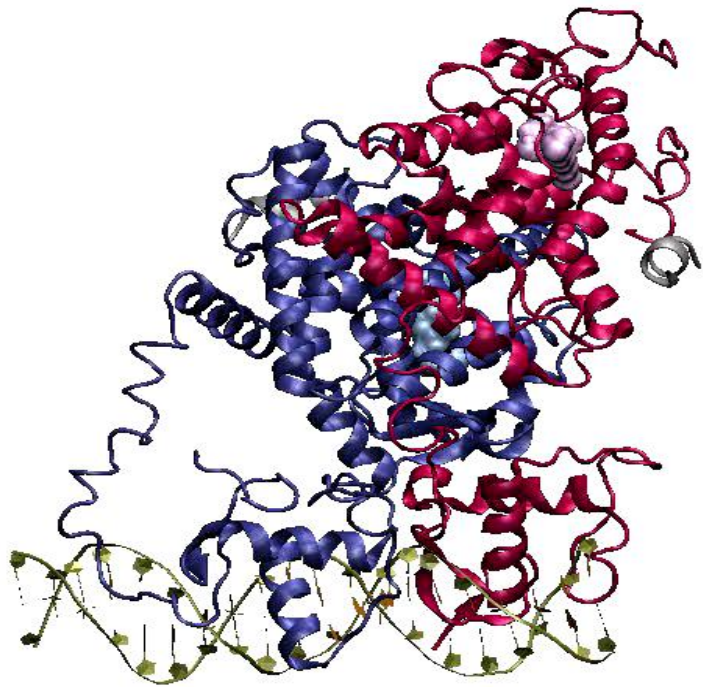
Full-length RXR/PPAR γ Complex: Dynamical Cross-Correlations

(Rastinejad, *Nature* 2008)



MD reveals correlated motions between DBD & H12 (~80 Å apart !)

K. Yamamoto et al, *Science* 2009:
DNA sequence affects NR activity --
“We therefore propose that DNA is a
sequence-specific allosteric ligand...”

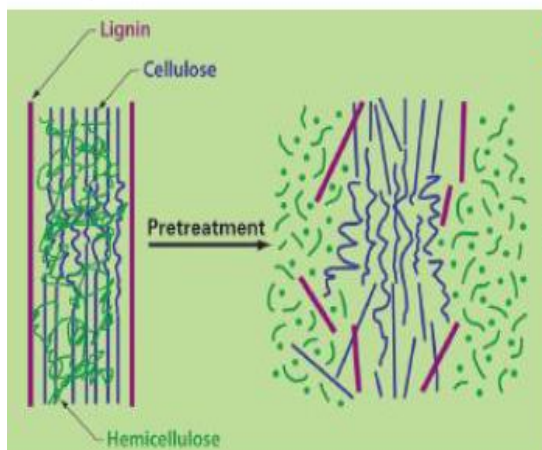


Glycosyl Hydrolases: Ethanol from Cellulose

Sugarcane Bagasse

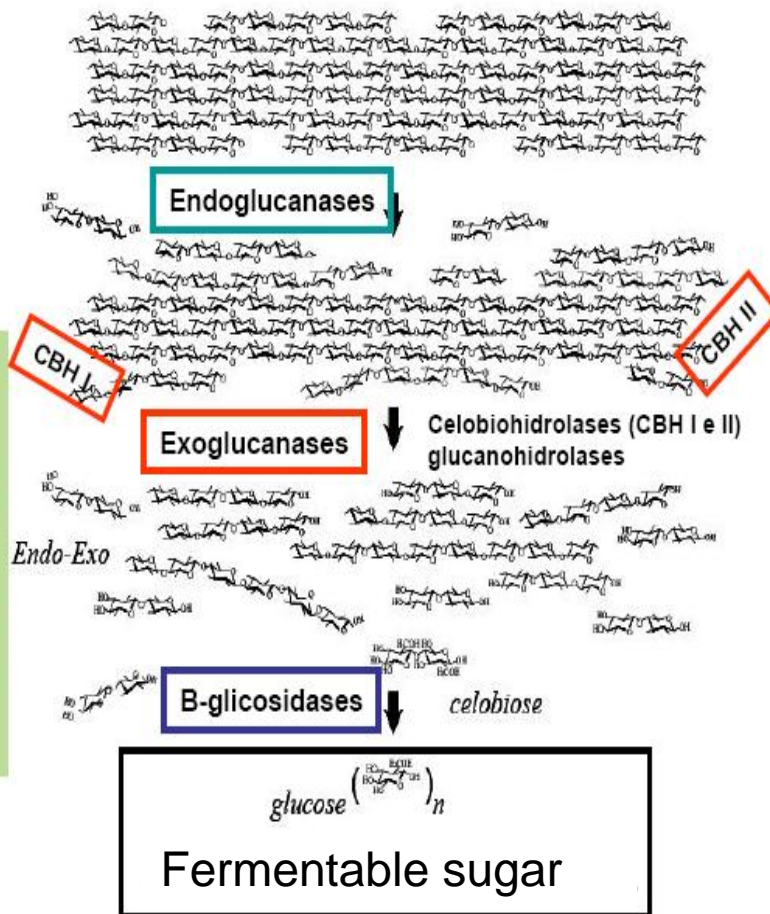


Parede celular



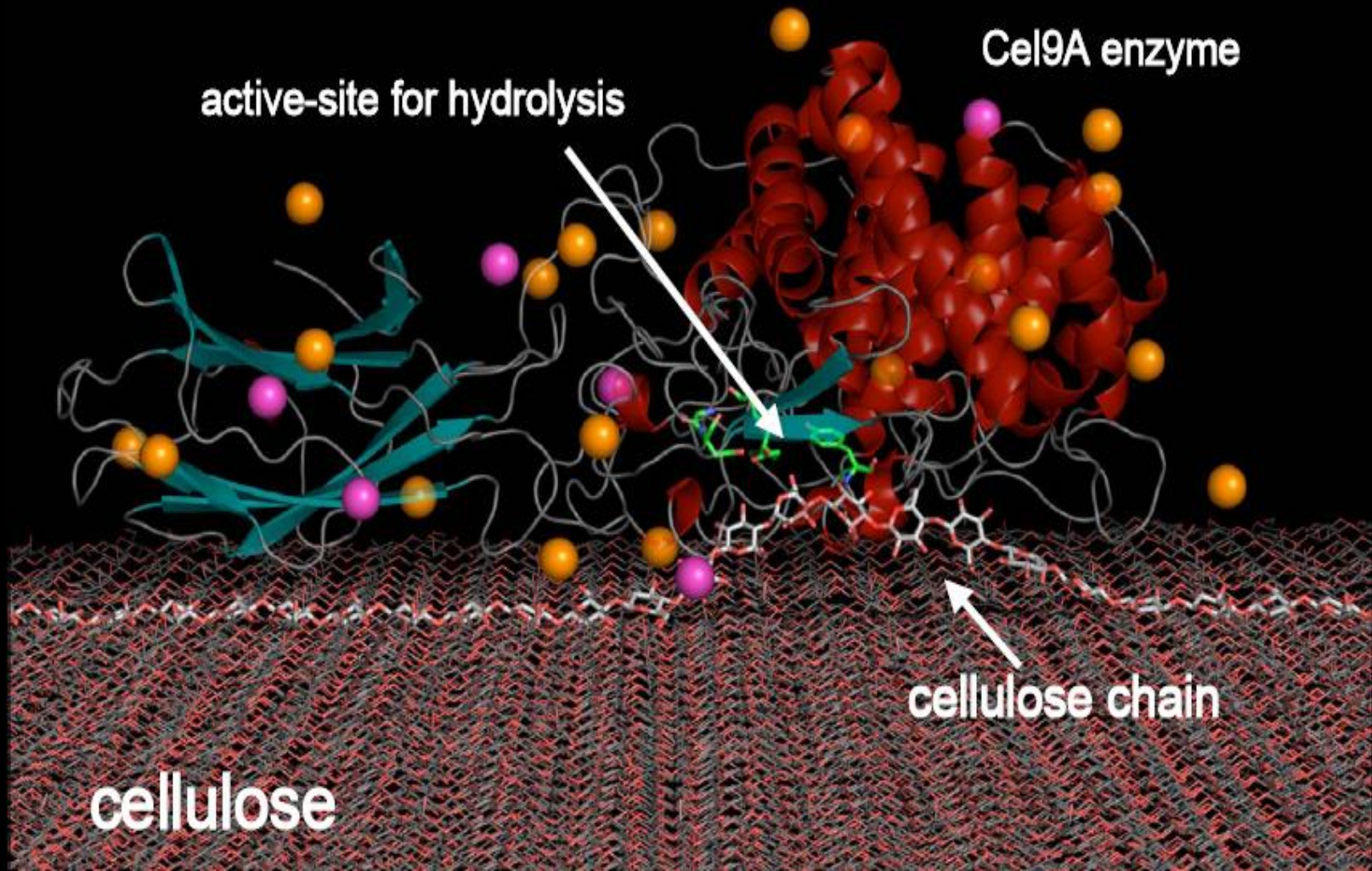
Lignina: 20%
 Hemicelulose: 28%
 → Celulose: 36%

→ Mechanism overview

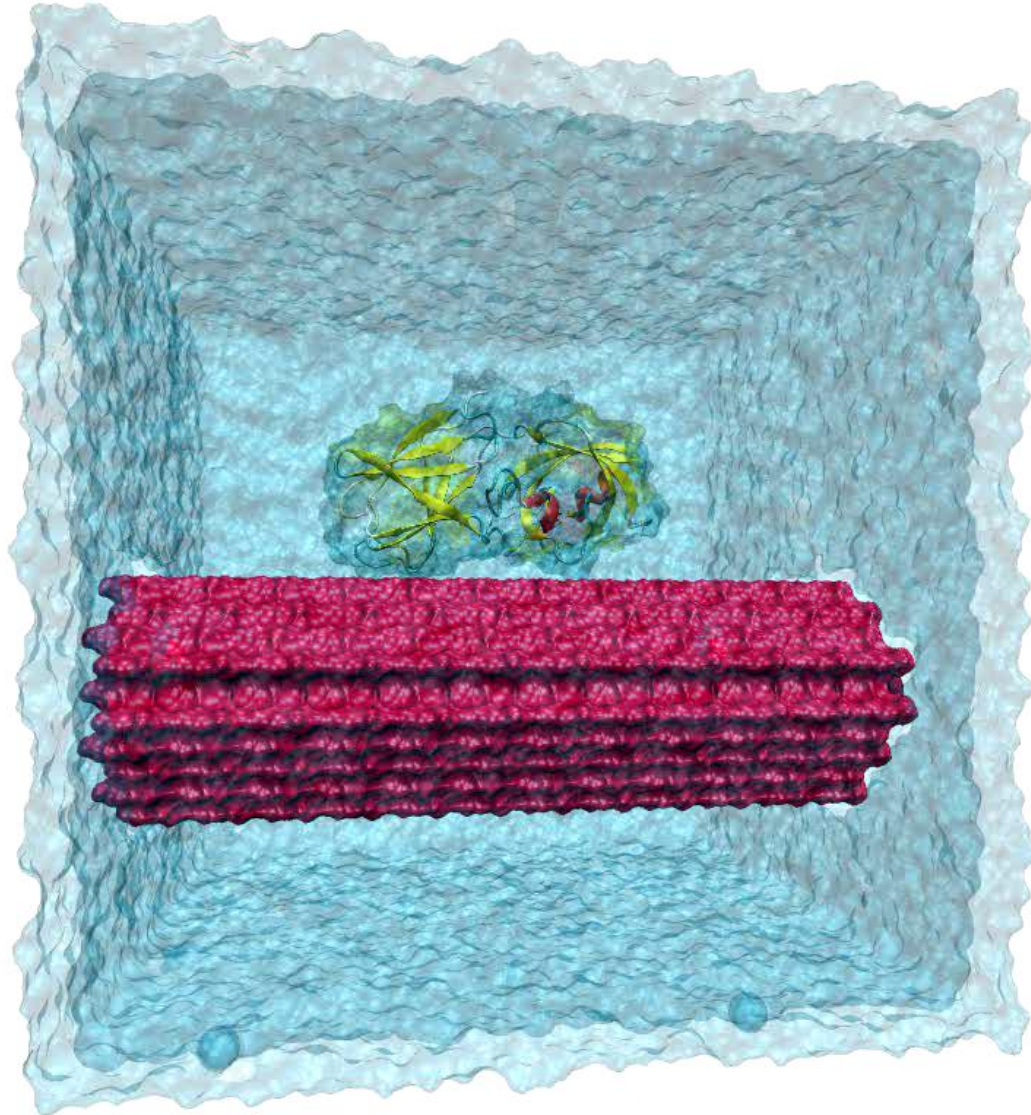


Goal: to develop efficient enzyme cocktails to render hydrolysis economically viable

Atomistic MD studies of GH

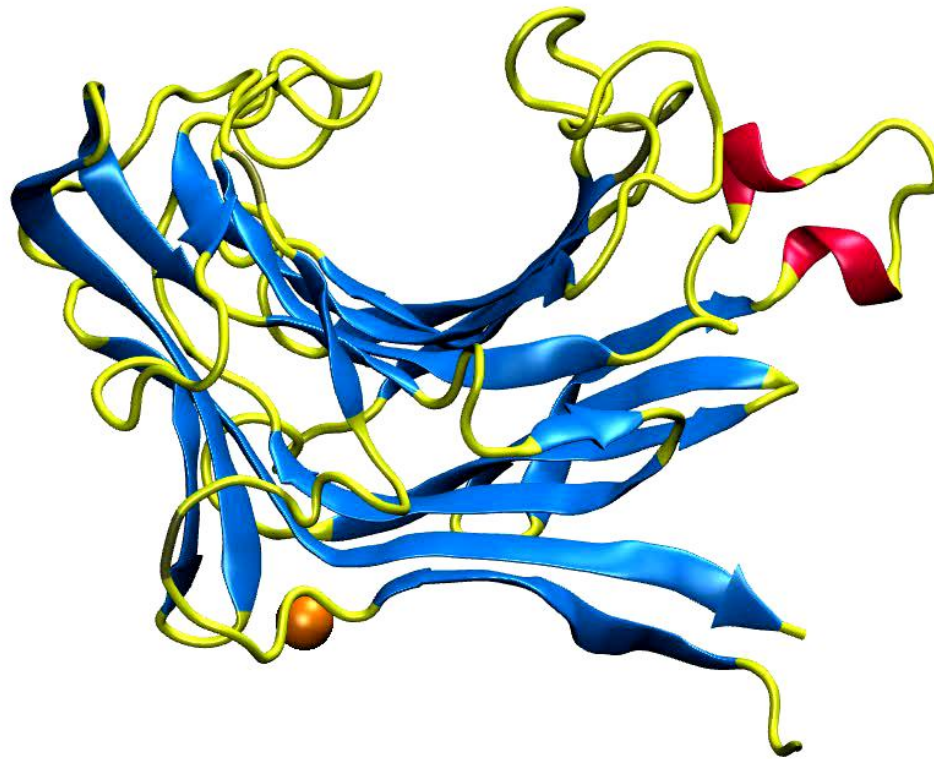


Binding to cellulose microfibrils (200,000 atoms)



Hyperthermophilic Laminarinase from *R. Marinus* (Family 16 GH; endoglucanase)

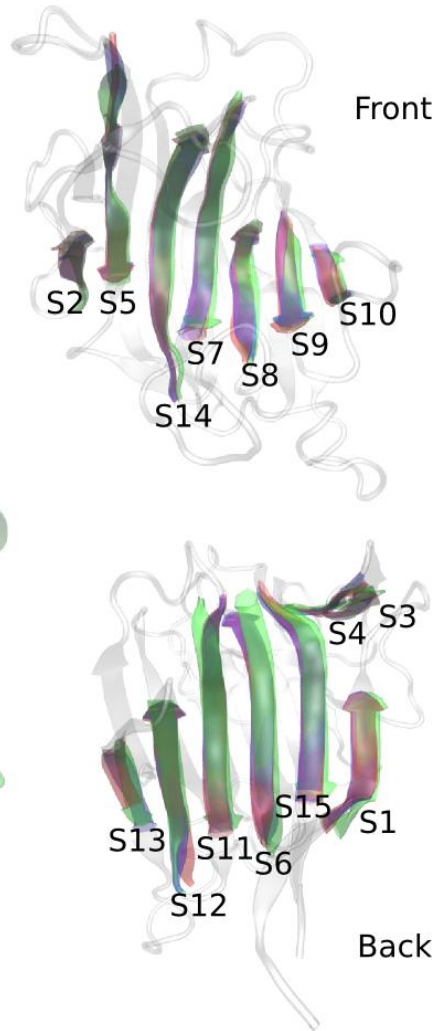
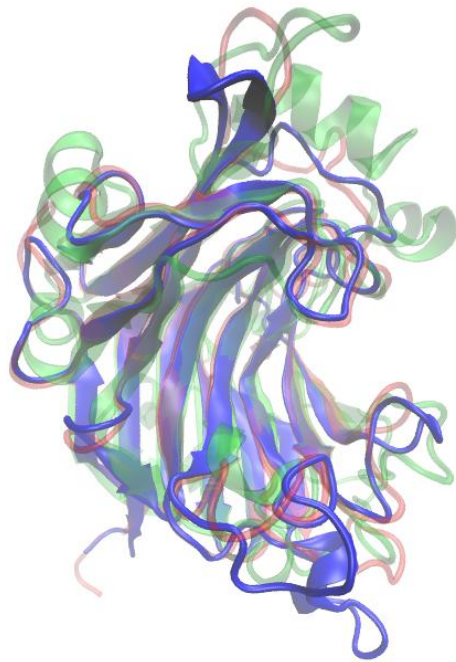
X-ray: 1.95 Å Resolution (I. Polikarpov)



Structural Homologs

2HYK (*Nocardiopsis* sp.) : Thermophile
2CL2 (*P. chrysosporium*): Mesophile

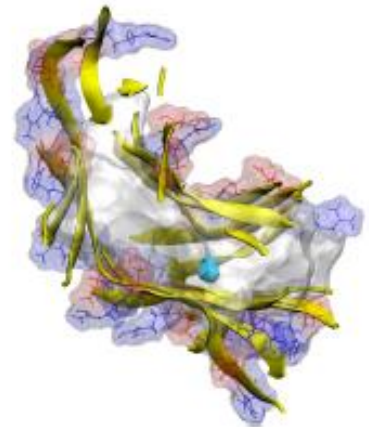
- RmLamR
- 2HYK
- 2CL2



Number of Salt Bridges:

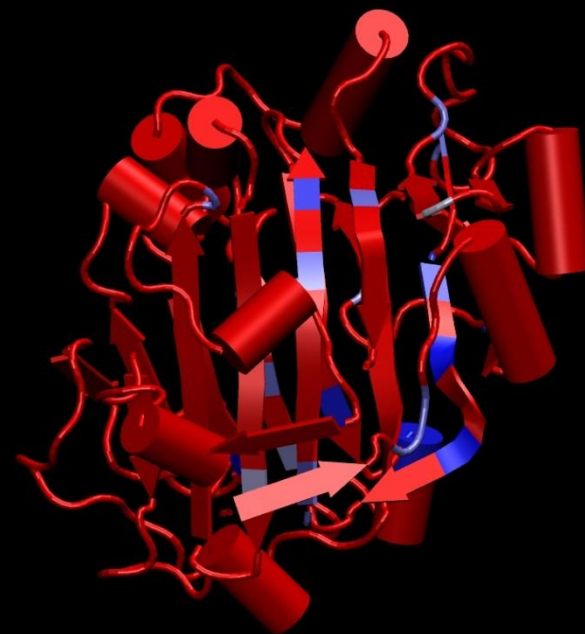
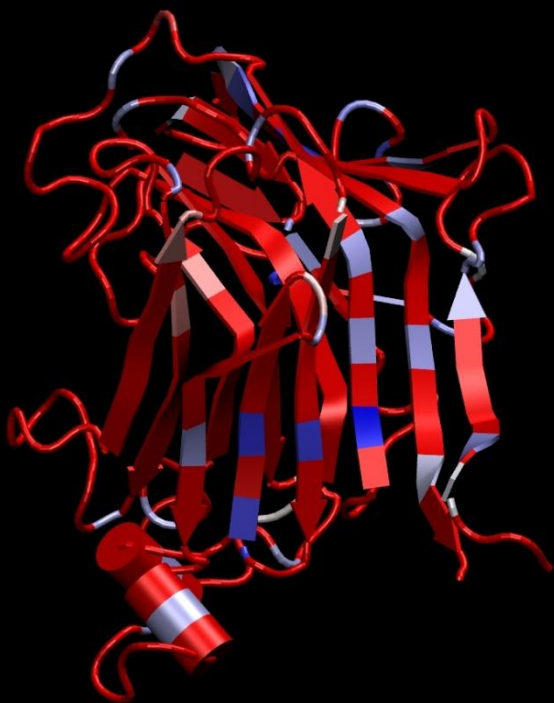
RmLamR	17 (22)
2HYK	7
2CL2	11

Charged residues interact equally well with water, so what is the enthalpic gain upon folding ?



HYDROPHOBIC CONTACTS AND SALT BRIDGES

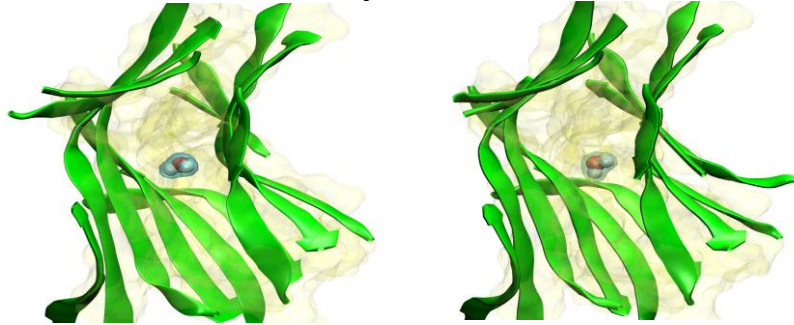
Enzyme	RmLamR		2HYK		2CL2	
Temp.	25 °C	90 °C	25 °C	90 °C	25 °C	90 °C
HC	166 ± 2	165 ± 3	150 ± 2	151 ± 3	153 ± 3	152 ± 4
SB	24 ± 3	24 ± 2	9 ± 1	10 ± 2	11 ± 1	11 ± 1



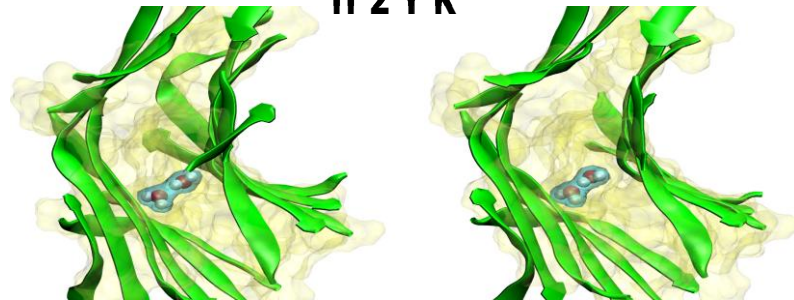
25 °C

90 °C

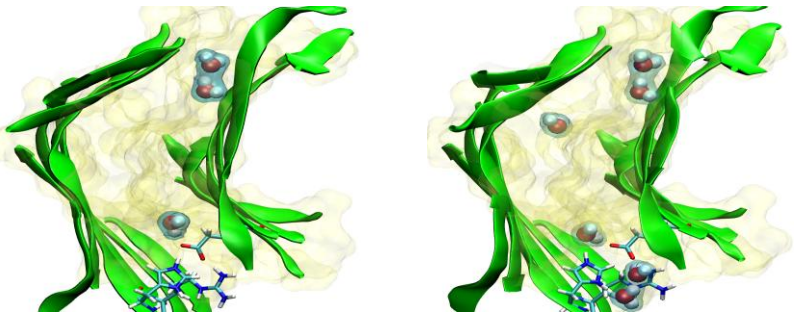
Lam R



H2YK



2CL2



**Molecular basis for
thermostability**

2CL2

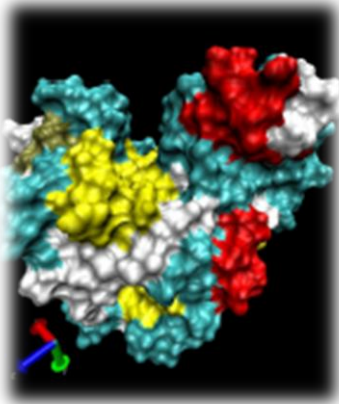
Charged residues in opposite leaflets interact via salt bridges.

More water molecules can diffuse into the hydrophobic core.

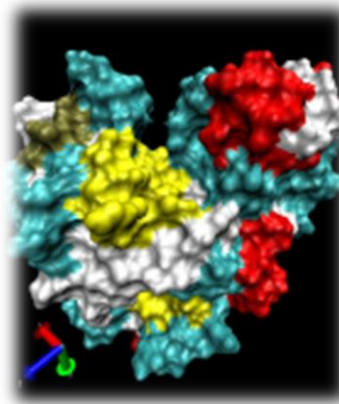
Molecular basis for thermophilicity

RmLamR: substrate binding channel is preserved

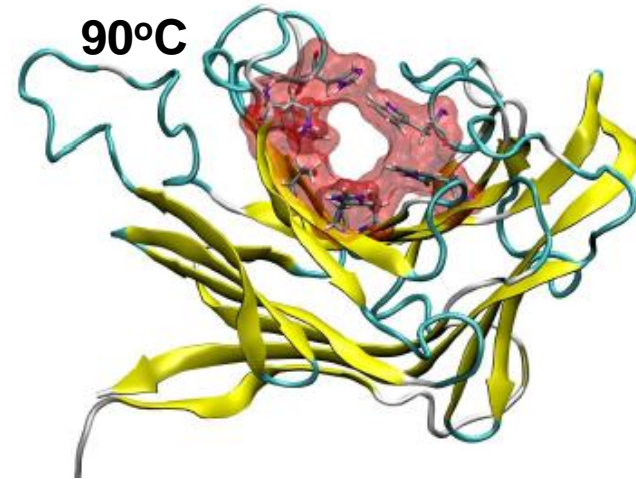
25°C



90°C

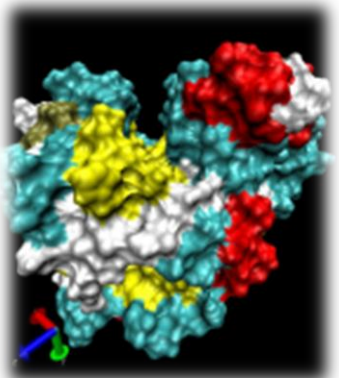


90°C

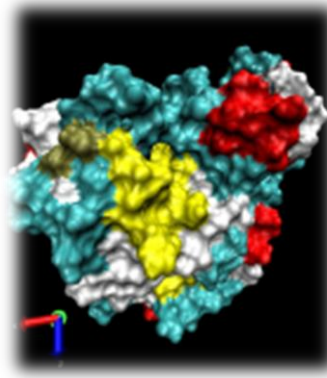


2CL2: substrate binding channel is obstructed at 90°C

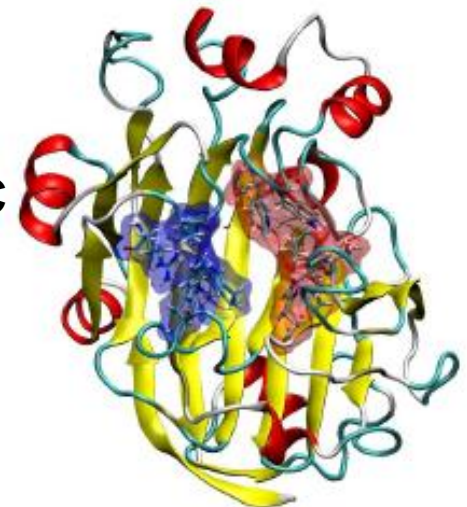
25°C



90°C



90°C



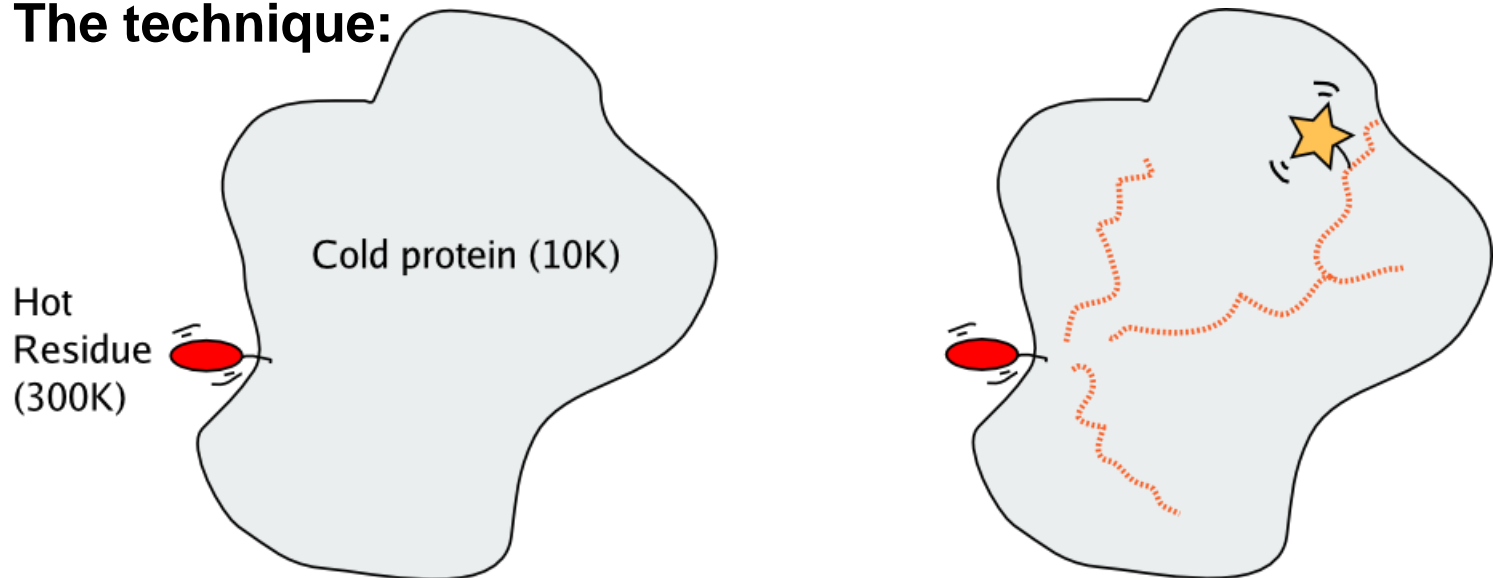
Intramolecular Vibrational Energy Transfer in Proteins (anisotropic thermal diffusion)

1. The method

1.1. Ota and Agard (2005):

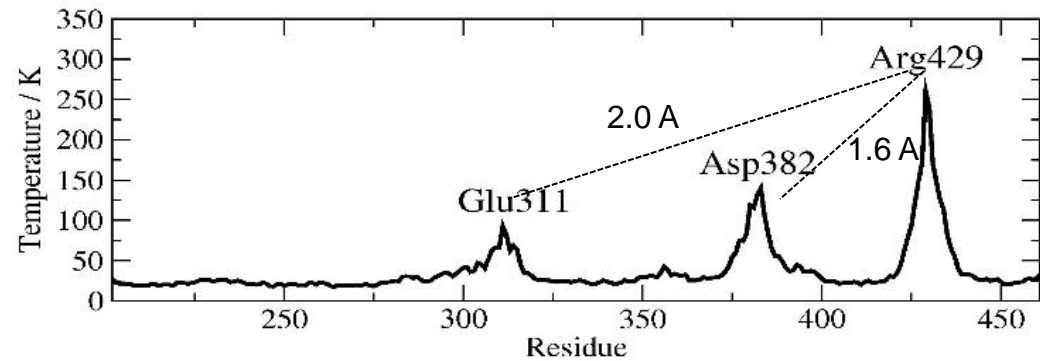
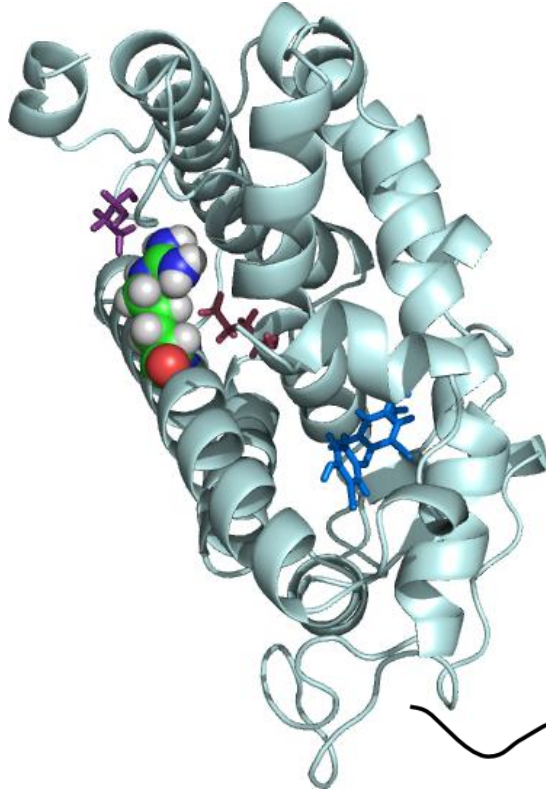
Thermal diffusion seems to explain networks of evolutionary connectivity (Ranganathan, Science 1999) - does not seem to be confirmed.

1.2. The technique:

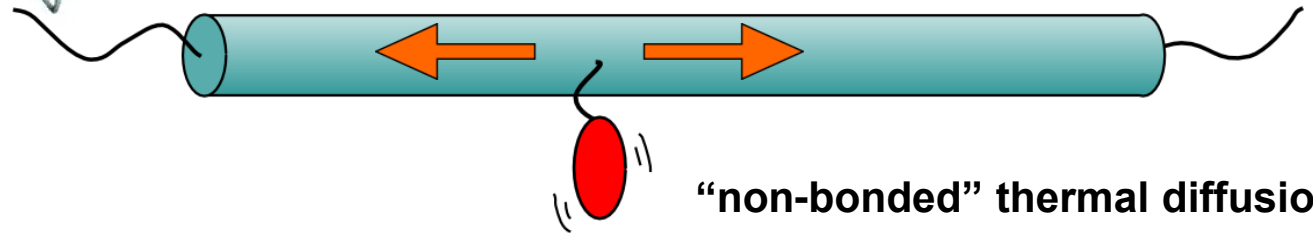


2. Mechanisms of thermal diffusion

Typical heating profile: Arg429

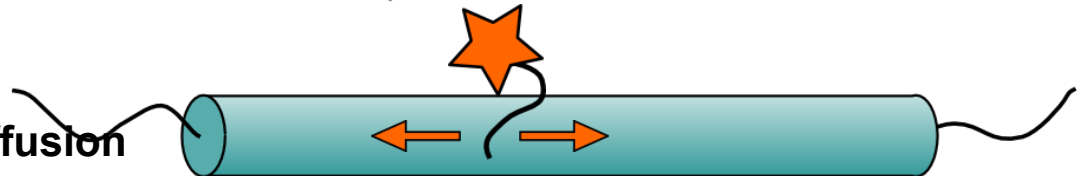


“bonded” thermal diffusion



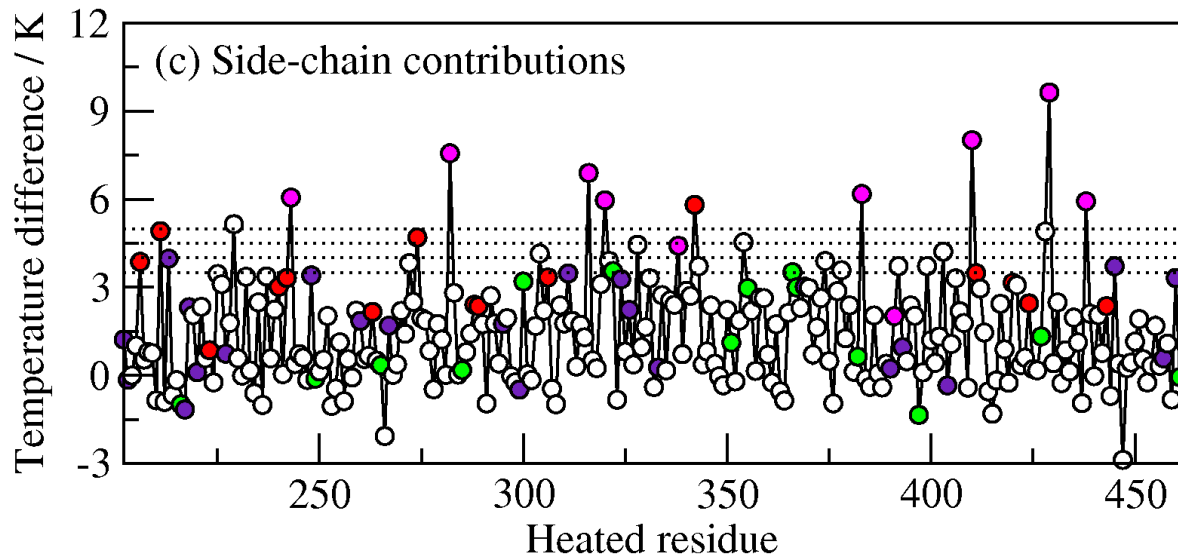
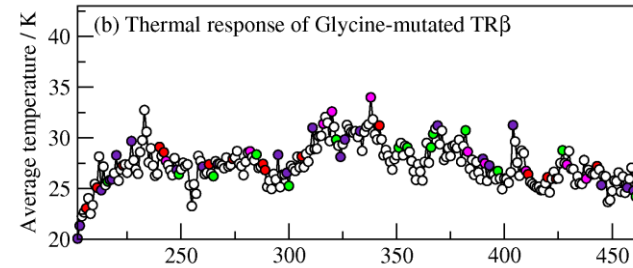
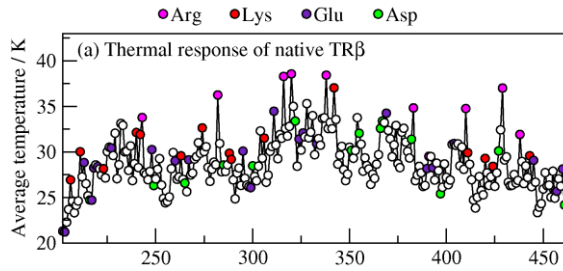
“non-bonded” thermal diffusion

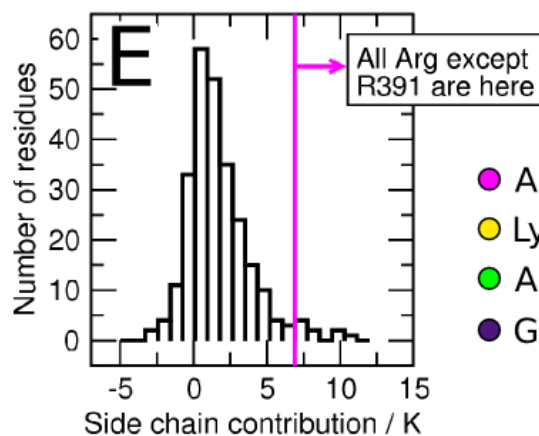
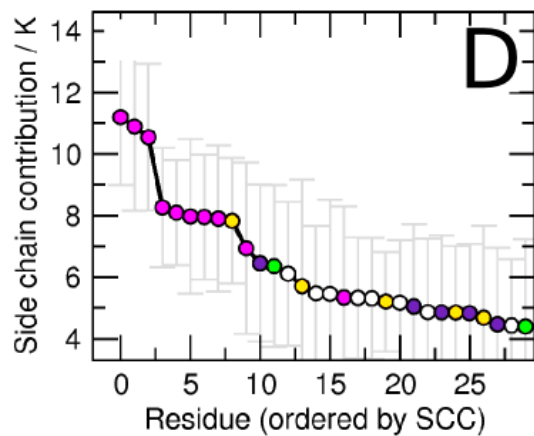
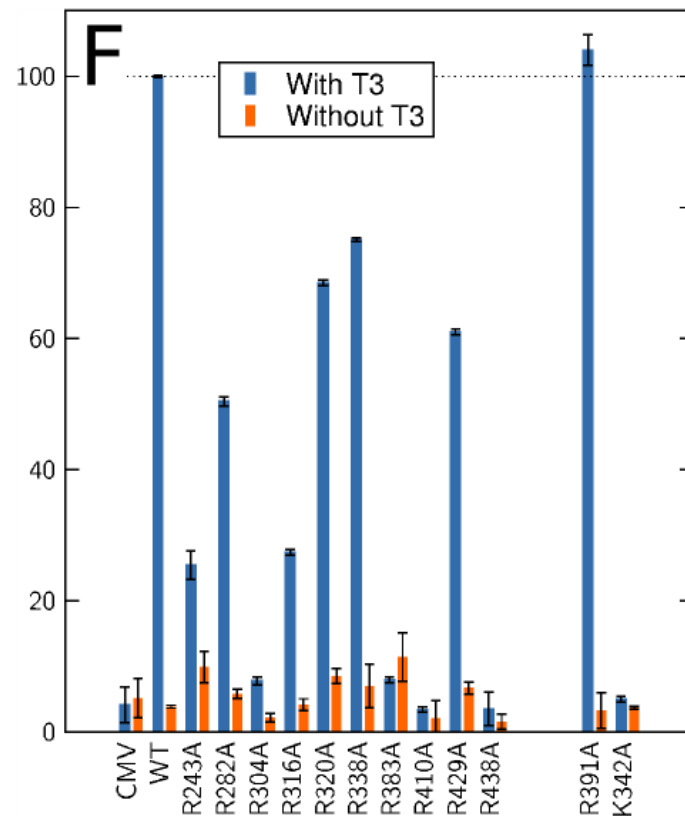
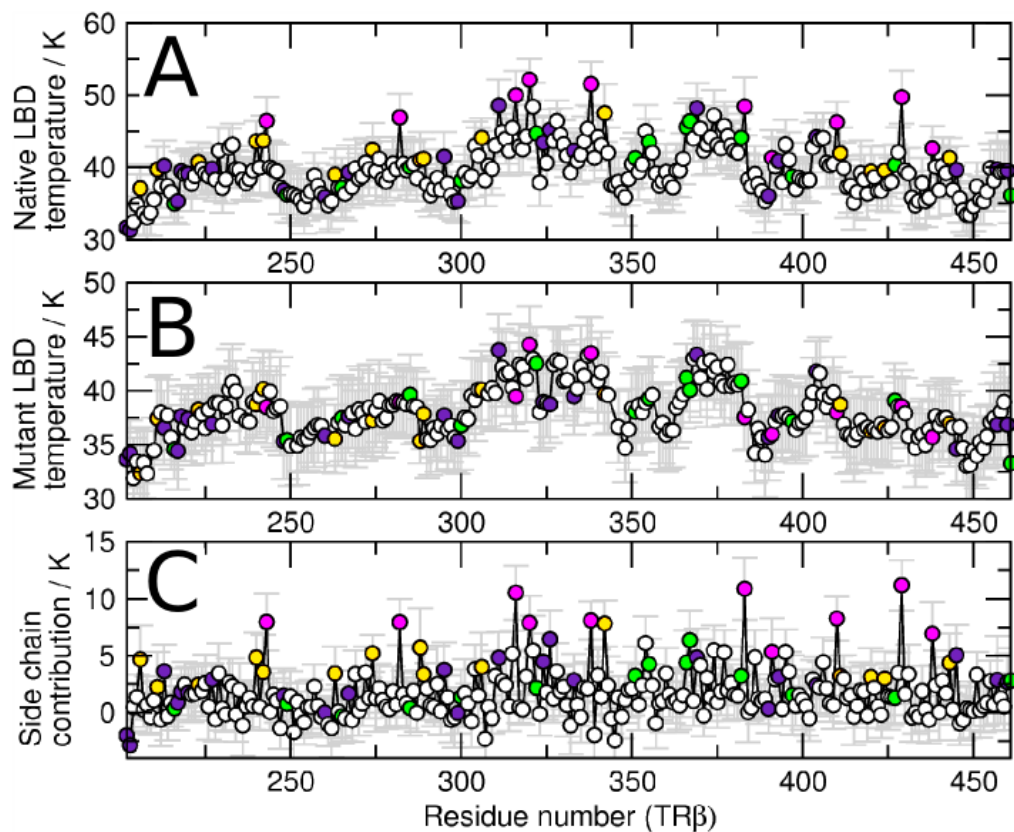
2^{ary} “bonded” thermal diffusion



2. Mechanisms of thermal diffusion

Side-chain contributions





- Arginine
- Lysine
- Aspartic acid
- Glutamic acid

Take home message

- MD simulations provide a powerful means of bridging the gap between protein structure and function
- MD may reveal unsuspected roles played by protein dynamics and water molecules (non-crystallographic) in protein function
- Experimentalists understand the results !

ACKNOWLEDGEMENTS

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